planar conformation (Ru-C4-O1-C11 dihedral angle 0.7°), consistent with π -donation from O1 to the carbene moiety and giving rise to a C4-O1 distance similar to those of other alkoxycarbenes.¹⁴ The short Ru—C4 distance of 2.002 (3) Å implies substantial π -donation from Ru to the carbene ligand, as found in Cp(CO)₂Mn and -Re systems.¹⁴ In common with arene dicarbonyl group 6 complexes,15 the carbene group in 2 is approximately perpendicular to the plane bisecting the CO-Ru-CO fragment; this is in contrast to Cp(CO)₂Mn-carbene complexes in which the M=C-O plane usually bisects the CO-M-CO angle, ¹⁶ in agreement with theory. ^{15a,17} This orientation may be sterically favored, although the appearance of a single CO resonance in the ¹³C NMR spectrum suggests rapid rotation about the Ru-C4 bond.

Complex 2 is relatively stable toward thermolysis, decomposing at 80 °C in THF with a half-life of 1.5 h in a sealed cell and forming the tricarbonyl precursor 1 as the only carbonyl-containing product. Reaction of 2 with an excess of 1-hexyne in refluxing toluene followed by oxidation of the reaction mixture generated the naphthoquinone 3 (Scheme I).¹⁸ This is the first observation of the benzannulation reaction¹⁹ for a group 8 metal-carbene complex,²⁰ occurring under conditions in which (MeCp)-(CO)₂MnC(Ph)(OMe) and Cp(CO)₂ReC(Ph)(OMe) are unreactive.16b

The isolation of 2 presages the synthesis of a new family of ruthenium (and perhaps other metal) carborane-stabilized carbenes with unusual properties, which can be tailored by varying the C_2B_1 ring substituents.²¹ We anticipate that these reagents will be relatively accessible, given the recent development of a facile synthesis of nido-R₂C₂B₄B₆ carboranes²² that we have employed on a multigram scale.

Acknowledgment. This work was supported by the National Science Foundation (Grant Nos. CHE 90-22713 to R.N.G. and 91-09000 to M.G.F.). We thank Dr. Kent Piepgrass for a sample of complex 1.

Supplementary Material Available: Listing of complete experimental details and tables of crystal structure data, thermal parameters, bond distances and angles, and mean planes (13 pages). Ordering information is given on any current masthead page.

(14) Schubert, U. In Transition Metal Carbene Complexes; Dötz, K. H., Fischer, H., Hofmann, P., Kreissl, F. R., Schubert, U., Weiss, K., Eds.; Verlag Chemie: Weinheim, Germany, 1983; Chapter 3.
(15) (a) Kostic, N. M.; Fenske, R. F. J. Am. Chem. Soc. 1982, 104, 3879.

(b) Schubert, U. J. Organomet. Chem. 1980, 185, 373.
(16) (a) Schubert, U. Organometallics 1982, 1, 1085. (b) Balzer, B. L.; Cazanoue, M.; Sabat, M.; Finn, M. G. Organometallics 1992, 11, 1759.
(17) Schilling, B. E. R.; Hoffman, R.; Lichtenberger, D. L. J. Am. Chem. Soc. 1979, 101, 585.

(18) A solution of 178 mg (0.286 mmol) of 2 in 8 mL of toluene was refluxed with 117 mg (1.43 mmol) of 1-hexyne for 3 h, and the reaction was judged complete by IR analysis. The flask was opened to air, the solvent was removed by evaporation, and the dark brown residue was taken up in ether and oxidized with 5.70 mL of 0.5 M ceric ammonium nitrate containing 0.1 M HNO₃ for 20 min. The organic phase was dried with MgSO₄ and the solvent was removed, affording an orange residue that was column-chromatographed in 10% ethyl acetate/petroleum ether to give 11 mg (0.051 mmol, 18%) of the naphthoquinone 3, which was identified by H NMR and comparison (NMR, capillary GLC) with an authentic sample prepared from (CO)₃CrC(Ph)(OMe) and 1-hexyne.¹⁹

(19) (a) Dötz, K. H. Angew. Chem., Int. Ed. Engl. 1975, 14, 644. (b) Wulff, W. D. In Advances in Metal-Organic Chemistry; Libeskind, L. S., Ed.; JAI Press Inc.: Greenwich, CT, 1989; Vol 1. (c) Wulff, W. D. In Comprehensive Organic Synthesis; Trost, B. M., Fleming, I., Eds.; Pergamon Press: New York, 1991; Vol. 5. (d) Bos, M. E.; Wulff, W. D.; Miller, R. A.; Chamberlin, S.; Brandvold, T. A. J. Am. Chem. Soc. 1991, 113, 9293.

(20) Reaction of tetracarbonyliron-carbene complexes with alkynes gives byrone products without annulation: Semmelhack, M. F.; Tamura, R.;

Schnatter, W.; Springer, J. J. Am. Chem. Soc. 1984, 106, 5363.
(21) (a) Davis, J. H., Jr.; Attwood, M. D.; Grimes, R. N. Organometallics 1990, 9, 1171. (b) Piepgrass, K. W.; Grimes, R. N. Organometallics 1992, 11, 2397. (c) Piepgrass, K. W.; Stockman, K. E.; Sabat, M.; Grimes, R. N. Organometallics 1992, 11, 2404.

(22) Cendrowski-Guillaume, S. M., Spencer, J. T. Organometallics 1992, 11, 969.

Intramolecular Benzannulation Reactions of Manganese Carbene Complexes

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The benzannulation reaction $^{1.2}$ of α,β -unsaturated Fischer carbene complexes with alkynes is largely restricted³ to complexes bearing the pentacarbonylchromium moiety. We have recently found that manganese carbene complexes activated with a titanoxy substituent undergo benzannulation with 1-hexyne in modest yields.⁴ Here we report that intramolecular benzannulation reactions of siloxycarbene compounds of manganese bearing tethered alkynes produce functionalized naphthoquinones in efficient fashion with complete regioselectivity under photochemical

Facile connection of the $MeCp(CO)_2Mn$ carbene (MeCp = η⁵-C₅H₄Me) and acetylenic alcohol fragments can be accomplished with the dimethylsilyl group, as shown in the general structure 1.5 The intramolecular benzannulation reaction of 1 is shown in eq 1.6

$$\begin{array}{c} R^{3}-C\equiv C+n \\ R^{3}-C\equiv C+n \\ (\eta^{5}-C_{5}H_{4}Me)(CO)_{2}Mn=C \\ \end{array} \xrightarrow{\begin{array}{c} 1. \ CH_{2}CI_{2}, \ 5 \ eq. \ PhCCPh. \\ 150W \ Xe \ arc. \\ \hline 2. \ aq. \ Ce^{1V} \\ \end{array}} \xrightarrow{\begin{array}{c} R^{2} \ OH \\ R^{3} \ () \ n \\ O = \\ \end{array}}$$

Preliminary results concerning the scope of the intramolecular benzannulation process under a standard set of conditions are summarized in Table I.7 As in intramolecular benzannulation reactions of chromium alkoxycarbene complexes,8 yields are low for terminal alkynes (entries 1, 5, 12, and 13) and a trimethylsilyl-substituted substrate (entry 9). In contrast, a variety of substituted propargylic and homopropargylic alcohols afford fair to good yields of naphthoquinones. No other products that can be obtained from alkyne insertion reactions of group 6 Fischer

- (1) Dötz, K. H. Angew. Chem., Int. Ed. Engl. 1975, 14, 644-646.
- (2) (a) Dötz, K. H.; Fischer, H.; Hofmann, P.; Kreissel, F. R.; Schubert, (2) (a) Dotz, K. H.; Fischer, H.; Hofmann, P.; Kreissel, F. R.; Schubert, U.; Weiss, K. Transition Metal Carbene Complexes; Verlag Chemie: Deerfield Beach, FL, 1984. (b) Wulff, W. D.; Tang, P. C.; Chan, K. S.; McCallum, J. S.; Yang, D. C.; Gilbertson, S. R. Tetrahedron 1985, 41, 5813-5832. (c) Wulff, W. D. In Advances in Metal-Organic Chemistry; Libeskind, L. S., Ed.; JAI Press Inc.: Greenwich, CT, 1989; Vol. 1. (d) Wulff, W. D. In Comprehensive Organic Synthesis; Trost, B. M., Fleming, I., Eds.; Pergamon Press: New York, 1991; Vol. 5. (e) Semmelhack, M. F.; Bozell, J. J.; Keller, L.; Sato, T.; Spiess, E. J.; Wulff, W.; Zask, A. Tetrahedron 1985, 41, 5803-5812
- (3) Wulff, W. D.; Gilbertson, S. R.; Springer, J. P. J. Am. Chem. Soc. 1986, 108, 520-522.
- (4) Balzer, B. L.; Cazanoue, M.; Sabat, M.; Finn, M. G. Organometallics **1992**, 11, 1759-1761.
- (5) All siloxycarbene complexes were prepared as in the following example, complex 1b: A solution of MeCpMn(CO), in dry Et₂O is treated with a slight excess of PhLi under inert atmosphere. ¹³ The lithium acylate, Li[MeCp-(CO)₂MnCOPh], precipitates as an air-sensitive ether solvate and can be used directly or recrystallized from Et2O. Under inert atmosphere, neat 2-butyn-1-ol is added dropwise with stirring to a 4-fold excess of neat SiMe₂Cl₂, and the volatiles are removed in vacuo to provide Me₂SiCl(OCH₂C=CMe). Complex 1b is then assembled in quantitative yield by the addition of the alkoxysilyl chloride to the acylate in CH₂Cl₂ or THF solution, followed by filtration through Celite to remove LiCl. All manganese siloxycarbene complexes were characterized by ¹H and ¹³C NMR, IR, and electronic spectroscopy (supplementary material).

(6) The intermolecular reaction of an analogous siloxycarbene complex MeCp(CO)₂MnC(Ph)(OSiMe₂O-n-Bu), with 1-hexyne provides a 15% yield of the corresponding quinone. See ref 4 for similar reactions.

(7) All quinone products were characterized by ¹H and ¹³C NMR, IR, and satisfactory elemental analyses (supplementary material).

(8) Semmelhack, M. F.; Bozell, J. J.; Keller, L.; Sato, T.; Spiess, E. J.; Wulff, W. D.; Zask, A. Tetrahedron 1985, 41, 5803-5812. (b) Wulff, W. D.; McCallum, J. S.; Kunng, F.-A. J. Am. Chem. Soc. 1988, 110, 7419-7434.

Table I. Intramolecular Benzannulation Reactions of Siloxycarbene Complexes^a

entry	complex	R1	R ²	R ³	n	quinone	yield (%) ^b
1	1a	Н	Н	Н	0	2a	30
2	1b	H	H	Me	0	2b	75
3	1c	H	H	Et	0	2c	51
4	1d	H	Н	Ph	0	2d	60
5	1e	Me	H	H	0	2e	20°
6	1f	Me	H	Me	0	2f	75
7	1g	Me	H	Et	0	2g	79
8	1h	OMe	H	Me	0	2h	48
9	1 i	Me	H	SiMe,	0	2 i	32^d
10	1i	Me	Et	Me	0	2j	51°
11	1k	Me	Н	Me	1	2k	56°
12	11	Me	H	Н	1	21	22
13	1m	Me	H	H	2	2m	23

^a Substituent positions refer to numbering scheme in eq 1. ^b Isolated yields by flash chromatography. ^c Acetylation in situ affords a 45% yield of the quinone acetate. ^d Compound 2i is unstable toward silica gel chromatography, giving diminished yields. ^e Values represent combined yields of hydroxyquinone and carbonyl product from overoxidation in workup.

carbene complexes, such as indenoid or cyclobutenone compounds,² are formed.

In every case in which the carbene aryl group is substituted (complexes 1e-m), a single isomer of the corresponding quinone (2e-k) is produced. Analogous regiocontrol in benzannulation reactions has previously been reported for tethered reactions of chromium alkoxycarbene complexes which provide tricyclic products. The present method extends the intramolecular strategy by providing bicyclic products with a convenient functional group for further elaboration.

Reaction conditions were optimized for complex 1f and include the following variables: (1) irradiation with a 150-W Xe arc lamp was found to be superior to Hg arc sources; (2) a glass filter with half-maximum absorption at 345 nm was found to protect the product from photochemical degradation¹⁰ and to insure selective irradiation of the lowest-energy electronic band of the siloxycarbene complexes at $\lambda_{max} \approx 360$ nm; and (3) the presence of 5 equiv of diphenylacetylene was found to roughly double the yield of naphthoquinone, without incorporation of the external alkyne. In all cases, diphenylacetylene can be recovered quantitatively. Carbon monoxide (1 atm), styrene, triphenylphosphine, dimethylacetylenedicarboxylate, and 2 equiv of diphenylacetylene were each found to be ineffective additives. Wulff and co-workers have shown the presence of an excess of alkyne to alter the course of benzannulation reactions of chromium, presumably by coordination to one or more intermediates in the multistep process.¹¹ Consistent with this rationale is our observation that the beneficial effect of 5 equiv of PhCCPh was eliminated in the presence of 1 atm of CO, but use of ¹³CO resulted in no incorporation of label in the product quinone.

The intramolecular nature of the reaction was demonstrated by a crossover experiment in which an equimolar mixture of complexes 1b and 1g gave only naphthoquinones 2b and 2g in 78% overall yield. In contrast to all benzannulation reactions of chromium and the intermolecular reaction of a manganese titanoxy system, 4 the intramolecular reactions of 1 require irradiation and could not be effected by heating. 12

Thus, the efficient and regioselective photochemical conversion of acetylenic alcohols to naphthoquinones has been accomplished, representing the first high-yield benzannulation chemistry of manganese carbene compounds. Studies concerning the scope, mechanism, and applications of this process will be reported in due course.

Acknowledgment is made to the National Science Foundation (CHE-9109000), the donors of the Petroleum Research Fund, administered by the American Chemical Society, and the Thomas F. and Kate Miller Jeffress Memorial Trust for the support of this research. We thank Dr. Michal Sabat for X-ray crystallographic analysis of the ketone derived from 2c, and Mr. Michael Gross for assistance in the regiochemical assignment of compound 2l.

Supplementary Material Available: Full details of experimental procedures and characterization of all compounds (15 pages). Ordering information is given on any current masthead page.

A Sequence-Specific Molecular Light Switch: Tethering of an Oligonucleotide to a Dipyridophenazine Complex of Ruthenium(II)

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There has been considerable interest in the development of new chemical methods to distinguish and detect nucleic acids with sequence specificity.¹ One focus of our laboratory has been on the application of ruthenium complexes^{2,3} as spectroscopic probes

⁽⁹⁾ Quantitative capillary GLC analysis of the crude reaction mixtures showed a single quinone in every case. Stereochemical assignments were made for representative cases by comparison with the isomeric producets produced in the reaction of $(CO)_5CrC(Ar)(OMe)$ with protected terminal alkynols, by observation of long-range $^1H^{-13}C$ NMR coupling, and by X-ray crystallography (supplementary material).

⁽¹⁰⁾ Both the isolated naphthoquinones and their organometallic precursors in the benzannulation reaction mixture were found to be unstable toward irradiation in the 300-345-nm range.

⁽¹¹⁾ Bos, M. E.; Wulff, W. D.; Miller, R. A.; Chamberlin, S.; Brandvold, T. A. J. Am. Chem. Soc. 1991, 113, 9293-9319.

⁽¹²⁾ Attempts to convert complex 1f to 2f by refluxing in CH₂Cl₂, THF, and toluene resulted in decomposition of the carbene complex to give no tractable organic products.

⁽¹³⁾ Fischer, E. O.; Maasböl, A. Chem. Ber. 1967, 100, 2445-2456.

^{*}Author to whom correspondence should be addressed.
(1) (a) Reid, T.; Baldini, A.; Rand, T. C.; Ward, D. C. Proc. Natl. Acad. Sci. U.S.A. 1992, 89, 1388. (b) Lichter, P.; Ward, D. C. Nature 1990, 345, 93. (c) Duvalvalentin, G.; Thuong, N. T.; Helene, C. Proc. Natl. Acad. Sci. U.S.A. 1992, 89, 504. (d) Sun, J. S.; Francois, J. C.; Montenaygarestier, T.; Saisonbehmoaras, T.; Roig, V. Proc. Natl. Acad. Sci. U.S.A. 1989, 86, 9198. (e) Moser, H. E.; Dervan, P. B. Science 1987, 238, 645. (f) Strobel, S. A.; Doucettestamm, L. A.; Riba, L.; Housman, D. E.; Dervan, P. B. Science 1991, 254, 1639. (g) Tizard, R.; Cate, R. L.; Ramachandran, K. L.; Wysk, M.; Voyta, J. C.; Murphy, O. J.; Bronstein, I. Proc. Natl. Acad. Sci. U.S.A. 1990, 87, 4514. (h) Murakami, A.; Tada, J.; Yamagata, K.; Takano, J. Nucleic Acids Res. 1989, 17, 5587. (i) Pollard-Knight, D.; Read, C. A.; Downes, M. J.; Howard, L. A.; Leadbetter, M. R.; Pheby, S. A.; McNaughton, E.; Syms, A. Ready, M. A. W. And Ricchem. 1990, 185, 84

Acids Res. 1989, 17, 3587. (1) Pollard-Knight, D.; Read, C. A.; Downes, M. J.; Howard, L. A.; Leadbetter, M. R.; Pheby, S. A.; McNaughton, E.; Syms, A.; Brady, M. A. W. Anal. Biochem. 1990, 185, 84. (2) (a) Barton, J. K.; Goldberg, J. M.; Kumar, C. V.; Turro, N. J. J. Am. Chem. Soc. 1986, 108, 2081. (b) Kumar, C. V.; Barton, J. K.; Turro, N. J. J. Am. Chem. Soc. 1985, 107, 5518. (c) Barton, J. K.; Basile, L. A.; Danishefsky, A.; Alexandrescu, A. Proc. Natl. Acad. Sci. U.S.A. 1984, 81, 1961. (d) Pyle, A. M.; Rehmann, J. P.; Meshoyrer, R.; Kumar, C. V.; Turro, N. J.; Barton, J. K.; J. Am. Chem. Soc. 1989, 111, 3051. (e) Turro, N. J.; Barton, J. K.; Tomalia, D. M. Acc. Chem. Res. 1991, 24, 332.

^{(3) (}a) Friedman, A. E.; Chambron, J.-C.; Sauvage, J.-P.; Turro, N. J.; Barton, J. K. J. Am. Chem. Soc. 1990, 112, 4960. (b) Friedman, A. E.; Kumar, C. V.; Turro, N. J.; Barton, J. K. Nucleic Acids Res. 1991, 19, 2595. (c) Hartshorn, R. M.; Barton, J. K. J. Am. Chem. Soc. 1992, 114, 5919. (d) Jenkins, Y.; Friedman, A. E.; Turro, N. J.; Barton, J. K. Biochemistry, in press.